

In-Band and Out-of-Band Emissions from Charge-Selected Sn Ions

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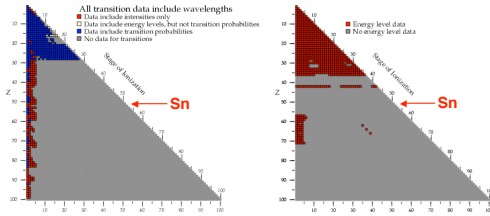
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How is the present status of atomic data ?

Spectroscopic data, namely transition wavelengths, transition probabilities, and energy levels of atoms in each ionic states, is **very far from complete** for the heavy elements such as **Sn (Z = 50)**.

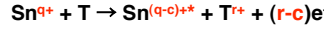


Elements and ionization stages for which the NIST Atomic Spectra Database contains data of transitions and energy levels.

<http://physics.nist.gov/>

What can we do for EUVL as atomic physicists ?

Development of **EUV** light sources for **Lithography** required spectroscopic data of multiply charged **Sn** ions. Hence, we measured EUV emission spectra from the individual charge states of Sn ions in **charge transfer collisions** with neutral gas targets to provide spectroscopic data.



↓ photon emission

$\text{Sn}^{(q-c)+}$

T : gas target atom

Principle of the charge exchange spectroscopy. The number of captured electron **c** is generally unity.

Setup for In-Band measurement

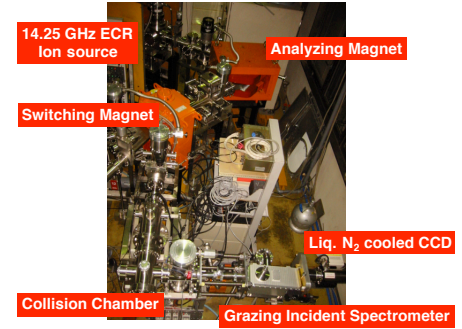
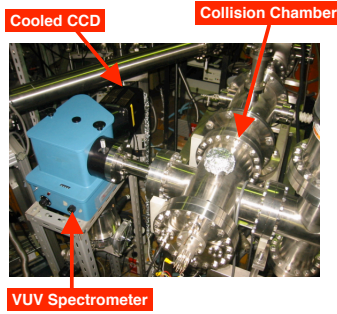
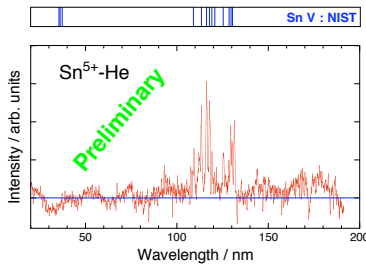


Photo of the experimental setup. Multiply charged ion facilities in Tokyo Metropolitan University.

Setup for Out-of-Band measurement



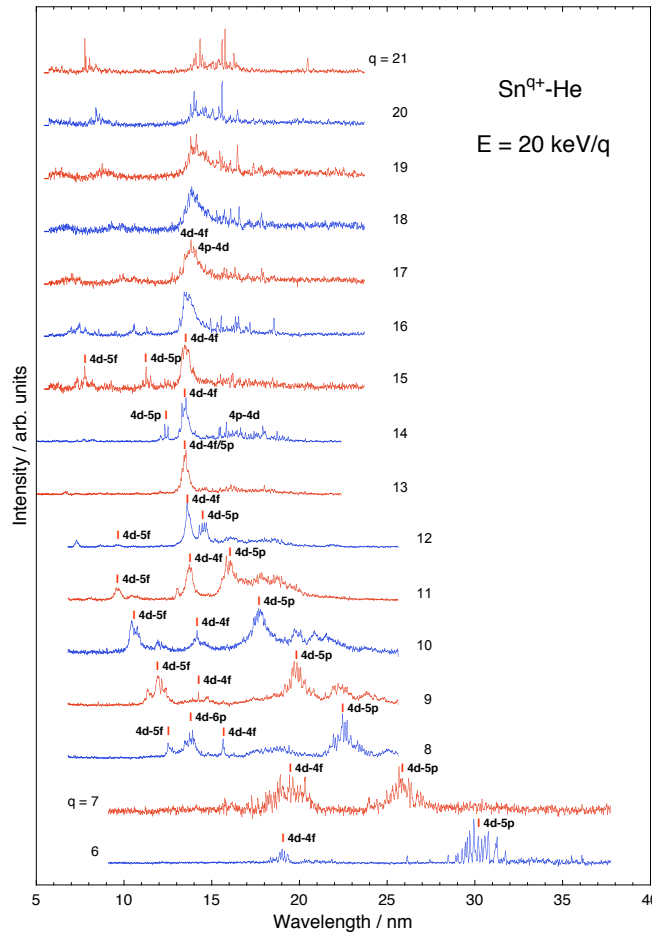
Out-Of-Band Emission



VUV emission spectra in collisions of Sn^{5+} ions with He gas at 20 keV/q.

Most of the observed lines from 108 to 130 nm are corresponding to the 5s-5p transitions of Sn V (Sn^{4+}). As the ground-state configuration of Sn V is $4d^{10} 1S_0$, the transitions between excited states are observed in this spectrum.

In-Band Emission



EUV emission spectra in collisions of Sn^{q+} ($6 \leq q \leq 21$) ions with He gas at 20 keV/q.

Sn^{q+}	$\text{Sn}^{(q-1)+}$	$\text{Sn}^{(q-1)+*}$	Resonance lines
$3d^{10}4s$	$3d^{10}4s^2$	$3d^{10}4s\text{nl}$	$4s\text{-nl}$
$4s^2$	$4s^24p$	$4s^2\text{nl}$	$4p\text{-nl}$
$4s^24p$	$4s^24p^2$	$4s^24p\text{nl}$	
$4s^24p^2$	$4s^24p^3$	$4s^24p^2\text{nl}$	
$4s^24p^3$	$4s^24p^4$	$4s^24p^3\text{nl}$	
$4s^24p^4$	$4s^24p^5$	$4s^24p^4\text{nl}$	
$4s^24p^5$	$4s^24p^6$	$4s^24p^5\text{nl}$	$4d\text{-nl}$
$4p^6$	$4p^64d$	$4p^6\text{nl}$	
$4p^64d$	$4p^64d^2$	$4p^64d\text{nl}$	
$4p^64d^2$	$4p^64d^3$	$4p^64d^2\text{nl}$	
$4p^64d^3$	$4p^64d^4$	$4p^64d^3\text{nl}$	
$4p^64d^4$	$4p^64d^5$	$4p^64d^4\text{nl}$	
$4p^64d^5$	$4p^64d^6$	$4p^64d^5\text{nl}$	
$4p^64d^6$	$4p^64d^7$	$4p^64d^6\text{nl}$	
$4p^64d^7$	$4p^64d^8$	$4p^64d^7\text{nl}$	$4d\text{-nl}$
$4p^64d^8$	$4p^64d^9$	$4p^64d^8\text{nl}$	

Configurations of ground states of Sn^{q+} and $\text{Sn}^{(q-1)+}$ ions, and singly excited $\text{Sn}^{(q-1)+}$ ions

Discussion

- The charge states of emitting Sn ions are considered to be $(q-1)^+$ because the single electron capture is generally the most dominant in collisions with He.
- VUV spectra corresponding to the out-of-band emissions can be observed with the same principle of the EUV measurements.
- EUV spectra consist of resolved lines and UTAs (unresolved transition arrays).
- Charge state dependence in higher ionization stages is not so strong, but spectra significantly depend on the charge state in lower ionization stages.
- Not only resonance lines, transitions between excited states have significant contribution in both the VUV and EUV emission spectra.